

This article was downloaded by: [217.201.65.168]

On: 06 June 2014, At: 12:07

Publisher: Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Natural Product Research: Formerly Natural Product Letters

Publication details, including instructions for authors and subscription information:

<http://www.tandfonline.com/loi/gnpl20>

Study on the antioxidant and antimicrobial activities of *Allium ursinum* L. pressurised-liquid extract

Dasha Sp. Mihaylova^a, Anna Lante^b, Federica Tinello^b & Albert I. Krastanov^a

^a Department of Biotechnology, University of Food Technologies, 26 Maritza Blvd., 4002 Plovdiv, Bulgaria

^b Department of Agronomy, Food, Natural Resources, Animals and Environment, Università di Padova, Viale dell'Università 16, Agripolis, 35020 Legnaro, Padova, Italy

Published online: 04 Jun 2014.

To cite this article: Dasha Sp. Mihaylova, Anna Lante, Federica Tinello & Albert I. Krastanov (2014): Study on the antioxidant and antimicrobial activities of *Allium ursinum* L. pressurised-liquid extract, *Natural Product Research: Formerly Natural Product Letters*, DOI: [10.1080/14786419.2014.923422](https://doi.org/10.1080/14786419.2014.923422)

To link to this article: <http://dx.doi.org/10.1080/14786419.2014.923422>

PLEASE SCROLL DOWN FOR ARTICLE

Taylor & Francis makes every effort to ensure the accuracy of all the information (the "Content") contained in the publications on our platform. However, Taylor & Francis, our agents, and our licensors make no representations or warranties whatsoever as to the accuracy, completeness, or suitability for any purpose of the Content. Any opinions and views expressed in this publication are the opinions and views of the authors, and are not the views of or endorsed by Taylor & Francis. The accuracy of the Content should not be relied upon and should be independently verified with primary sources of information. Taylor and Francis shall not be liable for any losses, actions, claims, proceedings, demands, costs, expenses, damages, and other liabilities whatsoever or howsoever caused arising directly or indirectly in connection with, in relation to or arising out of the use of the Content.

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden. Terms &

Conditions of access and use can be found at <http://www.tandfonline.com/page/terms-and-conditions>

Study on the antioxidant and antimicrobial activities of *Allium ursinum* L. pressurised-liquid extract

Dasha Sp. Mihaylova^{a*}, Anna Lante^b, Federica Tinello^b and Albert I. Krastanov^a

^aDepartment of Biotechnology, University of Food Technologies, 26 Maritza Blvd., 4002 Plovdiv, Bulgaria; ^bDepartment of Agronomy, Food, Natural Resources, Animals and Environment, Università di Padova, Viale dell'Università 16, Agripolis, 35020 Legnaro, Padova, Italy

(Received 20 February 2014; final version received 8 May 2014)

Allium ursinum L. is widely used as a spice as well as a traditional medicine. The aim of this work was to evaluate the antioxidant and antimicrobial activities (AMAs) of *A. ursinum* extract, obtained by pressurised-liquid extraction. Several reliable procedures such as 2,2-diphenyl-1-picrylhydrazyl, 2,2-azinobis-3-ethyl benxothiazoline-6-sulphonic acid, ferric-reducing antioxidant power assay and oxygen radical absorbance capacity assays were carried out. Vegetable oil stability was evaluated by using Rancimat test. Moreover, AMA was performed on different microorganisms. On the basis of the results obtained, it is confirmed that the *A. ursinum* extract could be used as a natural ingredient in food and/or pharmaceutical industries.

Keywords: *A. ursinum*; polyphenols; antioxidant activity; antimicrobial activity; Rancimat test

1. Introduction

Lipid autoxidation is a free radical chain reaction, leading to increase in reactive radicals and hydroxides, which initiate further transmutations (Frankel 1985; Min & Boff 2002). The use of antioxidants for the control of oxidation process is desirable from technological and nutritional points of view. Recently, the research is aimed to isolate and characterise bioactive compounds from natural sources which possess antioxidant ability and are safe for human consumption. The species belonging to the *Allium* family have been used for a long time as a remedy for the prevention and treatment of certain diseases (Bagheri et al. 2011).

Allium ursinum L., which is known as 'wild garlic', 'ramsons', 'bear's garlic', is a wild relative of chives native to Europe. It is widely used as a spice as well as medicinal plant (Janeczko & Sobolewska 1995). In medieval medicine, the leaves of *A. ursinum* were used as a therapy for cardiovascular diseases (Richter 1999). Several biological activities of this plant, such as antioxidative (Wu et al. 2009), cytostatic (Saniewska & Zuradzka 2001) and antimicrobial (Ivanova et al. 2009; Sapunjeva et al. 2012), were reported.

The extraction process is the first crucial step in the preparation of plant formulations. Handa et al. (2008) described the pressurised-liquid extraction (PLE) technique as the conventional method used to recover natural products. PLE is a sustainable and green technique to extract bioactive compounds from food by-products of plant origin using aqueous ethanol or water as food-grade solvents (Wijngaard et al. 2012). This extraction technique is gaining popularity because it requires less time and a lower consumption of organic solvents than conventional techniques (Mendiola et al. 2007). In addition, PLE can be used to extract more polar

*Corresponding author. Email: dashamihaylova@yahoo.com

compounds such as polyphenols recognised for many health benefits due to their antioxidant activity (AOA). However, their extracting efficiency depends mainly on temperature, type of solute and extraction medium (Wijngaard et al. 2012).

This work focuses on the characterisation of AOA and antimicrobial activities (AMAs) of an *A. ursinum* extract obtained for the first time by PLE technique in mild conditions in order to apply it in the food and/or the pharmaceutical industries.

2. Results and discussion

2.1 Total polyphenol content

The total polyphenol content (TPC; Table 1) of the investigated extract was 60.92 ± 0.32 mg gallic acid equivalent (GAE) 100 g^{-1} fresh weight (FW) of leaves. Sapunjieva et al. (2012) reported a higher concentration such as 145 mg GAE 100 g^{-1} FW using the 96% ethanol as solvent in heat-reflux, probably due to the different extraction technique and the raw material. In this respect, Pârnu, Toiu, et al. (2010) reported qualitative and quantitative differences in the polyphenolic composition of *Allium* species. Moreover, Schmitt et al. (2005) showed that the concentration of volatile precursors of *A. ursinum* depends on the harvesting time.

2.2 AOA of *A. ursinum* extract

Recently, polyphenolic content and AOA have been successfully correlated by several studies (Tepe et al. 2006; Mihaylova et al. 2013). However, Štajner, Igić, et al. (2008) and Štajner, Popovic, et al. (2008) showed that the investigated *Allium* species exhibited antioxidant properties in all plant organs, the highest having been observed in the leaves. The AOA of *A. ursinum* pressurised-liquid extract was evaluated with different assays in order to better understand the mechanism of action of an antioxidant (Jimenez-Alvarez et al. 2008). The results of DPPH[•] and ABTS^{•+} assays were, respectively, 0.25 ± 0.01 and 2.65 ± 0.04 μM Trolox equivalents (TE) g^{-1} FW (Table 1). In comparison, Sapunjieva et al. (2012) reported different DPPH[•] and ABTS^{•+} values, respectively, 1.07 and $16.34 \mu\text{M TE g}^{-1}$ FW with 96% ethanol probably due to the different origin of the plant samples and the extraction technique. *A. ursinum* extract showed a ferric-reducing antioxidant power (FRAP) assay value of $2.02 \pm 0.01 \mu\text{M TE g}^{-1}$ FW and a oxygen radical absorbance capacity (ORAC) value of $60.35 \pm 0.98 \mu\text{M TE g}^{-1}$ FW as reported for lovage and chilli pepper (Číž et al. 2010). However, it is very hard to evaluate deeply the AOAs of an extract on the basis of *in vitro* assays only. Taking into account that *A. ursinum* was employed for a long time in the folk medicine without data concerning its toxicity, other *in vivo* assays to assess the possibility of technological application may be more useful.

2.3 Vegetable oil stability using *A. ursinum* extract and AMA of the extract

Edible fats and fat containing products undergo oxidation, both during production and storage, causing a sequence of unfavourable changes. The enrichment of lipids with plant polyphenols may improve their oxidative stability and positively influence human health (Kozłowska et al.

Table 1. Total polyphenol content (mg GAE 100 g^{-1} FW) and antioxidant activity ($\mu\text{M TE g}^{-1}$ FW) of *A. ursinum* extract.

Sample/assay	TPC	TEAC _{DPPH}	TEAC _{ABTS+}	FRAP	ORAC
PLE	60.92 ± 0.32	0.25 ± 0.01	2.65 ± 0.04	2.02 ± 0.01	60.35 ± 0.98

Note: The data are expressed as mean \pm SD ($n = 3$).

2012). In this respect, many authors studied the stability of vegetable oils using various plant extracts (Ivanova et al. 2009; Proestos et al. 2013) and essential oils (Quiroga et al. 2013). The results on the oxidative stability of corn, peanut, soybean and sunflower oils emulsified with *A. ursinum* PLE reported statistically significant differences (Table 2). The peanut oil emulsion exhibited the highest value of AOA index (AAI, 1.69), suggesting an improved oxidative stability (69.85%). On the contrary, the soybean oil emulsion had the lowest AAI value (1.40), indicating a worse oxidative stability (39.52%). In addition, *A. ursinum* PLE exhibited a better result than red chicory extract tested in a previous study (Lante et al. 2011).

The results of antimicrobial screening given in Table 3 did not show any significant activity of the *A. ursinum* PLE against the tested organisms (TOs). However, higher AMAs were obtained towards *Staphylococcus aureus* (12 and 10 mm inhibition zones) and towards *Aspergillus niger* (6 mm inhibition zones). Other studies showed that the antibacterial and antifungal effects of *Allium* species depend mostly on the pathogenic species and on the type of plant extract (Sobolewska et al. 2006; Ivanova et al. 2009; Pârnu, Pârnu, et al. 2010, Sapunjieva et al. 2012).

3. Experimental

3.1 Plant material and extraction procedure

For each experiment, about 100 g of vegetable was randomly sampled from 500 g of *A. ursinum* L. (leaves and flowers) harvested in 2012 in Veneto region (Italy) during the flowering period and stored in a freezer until use. The sample was cut into small pieces and extracted as described by Mihaylova et al. (2014) with 500 mL of ethanol/water solution (85:15, v/v) containing 0.1 M HCl (pH 3.5), previously deoxygenated by flushing with nitrogen (Rossetto et al. 2005). The volume of each sample was measured to calculate the extract concentration (g L^{-1}), the extraction rate was 0.1754. Then, PLE was split up in dark glass bottles without headspace and stored at -20°C .

3.2 Determination of TPC

The TPC was analysed using the Folin–Ciocalteu method of Kujala et al. (2000) with some modifications (Mihaylova et al. 2014). Results were expressed as mg GAE per g FW.

3.3 DPPH radical-scavenging activity assay

The DPPH radical-scavenging activity was determined following the method of Brand-Williams et al. (1995) slightly modified by Mihaylova et al. (2014). The results were expressed as TE antioxidant capacity – TEAC value ($\mu\text{M TE g}^{-1}$ FW).

Table 2. Oxidative stability of *A. ursinum* extract oil emulsions measured by Rancimat® test.

Sample	IT (h)	AAI
Corn oil	$7.71 \pm 0.30\text{a}$	1.00
Corn oil + extract	$11.37 \pm 0.21\text{a}$	1.47
Peanut oil	$7.43 \pm 0.20\text{b}$	1.00
Peanut oil + extract	$12.62 \pm 0.27\text{b}$	1.69
Soybean oil	$5.39 \pm 0.32\text{c}$	1.00
Soybean oil + extract	$7.52 \pm 0.15\text{c}$	1.40
Sunflower oil	$3.50 \pm 0.49\text{d}$	1.00
Sunflower oil + extract	$5.23 \pm 0.74\text{d}$	1.50

Note: a, b, c, d: $P < 0.05$.

Table 3. AMA of *A. ursinum* extract.

Test organism/sample	Inhibition zone (mm)		Concentration of TO, CFU mL ⁻¹ in the media
	Control	Sample	
<i>Escherichia coli</i> ATCC 25922	–	Bacteria growth	5.0×10^{10}
<i>Klebsiella</i>	–; 1	1; 1	9.6×10^{10}
<i>Salmonella enterica</i> subsp. <i>enterica</i> ATCC BAA-2162	–	–	1.0×10^{11}
<i>Pseudomonas aeruginosa</i> ATCC 9027	–	–	1.27×10^{12}
<i>S. aureus</i> ATCC 25093	–	12; 10	5.0×10^{11}
<i>Listeria monocytogenes</i>	–	–	2.5×10^6
<i>Bacillus subtilis</i>	–	–	1.0×10^5
<i>Saccharomyces cerevisiae</i>	–	–	8.0×10^4
<i>A. niger</i>	–	6; 6	1.0×10^5
<i>Rhizopus</i>	–	–	1.0×10^5

Note: (–) No activity detected. The data are results of two parallel determinations. CFU, colony-forming units.

3.4 ABTS radical cation decolourisation assay

The scavenging activity of the extract against radical cation (ABTS⁺) was estimated according to Re et al. (1999). The results were expressed as TEAC ($\mu\text{M TE g}^{-1}$ FW).

3.5 FRAP assay

The FRAP assay was carried out according to the procedure of Benzie and Strani (1996) with some modifications (Mihaylova et al. 2014). The results were expressed as $\mu\text{M TE g}^{-1}$ FW.

3.6 ORAC assay

The ORAC assay measures the antioxidant-scavenging function against peroxy radical induced by (2,2'-azobis(2-amidinopropane) dihydrochloride at 37°C. The loss of fluorescence of fluorescein indicates the extent of damage due to its reaction with the peroxy radical (Ou et al. 2001). The ORAC values were expressed as $\mu\text{M TE g}^{-1}$ FW.

3.7 Determination of the oxidative stability

Oxidative stability is expressed in hours and determined according to AOCS official method Cd 12b (AOCS 1992), using a Rancimat apparatus (Metrohm, model 743, Herisau, Switzerland). The determination was made at 110°C with a 20 L h⁻¹ air flow, using 3 g, respectively, of corn, peanut, sunflower and soybean oils with *A. ursinum* PLE dissolved in a water/ethanol emulsions and 60 mL of distilled water in flasks containing electrodes. The volatile decomposition products were detected as reported by Lante et al. (2011).

The induction time (IT) is the time (h) at the break point of the two extrapolated straight parts of the curve obtained using the Rancimat® apparatus. The AAI was calculated by the following equation:

$$\text{AAI} = \frac{\text{IT of oil emulsions with extract}}{\text{IT of oil emulsions without extract}}$$

3.8 Determination of AMA

The AMA of the examined extracts was analysed by the agar-well diffusion method as described by Mihaylova et al. (2014). All the plates were examined for any zones of growth inhibition, and the diameters of the zones were measured in mm.

3.9 Statistical analysis

All measurements (except AMA – in duplicate) were carried out in triplicates. The results were expressed as mean \pm SD and statistically analysed using MS-Excel software.

Rancimat data were subjected to analysis of variance. The significance of differences was determined using Tukey's multiple-range test ($P < 0.05$) using the CoHort software package (CoHort Software, Monterey, CA, USA).

4. Conclusions

On the basis of the results obtained, *A. ursinum* PLE exerted significant AOA and improved vegetable oil stability. Further studies will focus on optimising the extraction technique and the polyphenol concentration, in order to apply *A. ursinum* extract as a non-toxic ingredient in food and/or pharmaceutical industries.

Acknowledgements

This work was supported by the 'Science and Business' Program at the Ministry of Education and Science, Bulgaria, BG051PO001/3.3-05, co-financed by the European Social Fund. The authors thank Stefania Zannoni for technical assistance.

References

- American Oil Chemists Society [AOCS]. 1992. Official methods and recommended practices of the American oil chemists' society. 5th ed. Champaign, IL: AOCS; p. Cd 12b.
- Bagheri F, Gol A, Dabiri S, Javadi A. 2011. Preventive effect of garlic juice on renal reperfusion injury. *Iran J Kidney Dis.* 5:194–200.
- Benzie IFF, Strani JJ. 1996. The ferric reducing ability of plasma (FRAP) as a measure of 'antioxidant power': the FRAP assay. *Anal Biochem.* 239:70–76.
- Brand-Williams W, Cuvelier ME, Berset C. 1995. Use of a free radical method to evaluate antioxidant activity. *LWT Food Sci Technol.* 28:25–30.
- Číž M, Čížová H, Denev P, Kratchanova M, Slavov A, Lojek A. 2010. Different methods for control and comparison of the antioxidant properties of vegetables. *Food Control.* 21:518–523.
- Frankel EN. 1985. Chemistry of autoxidation: mechanism, products and flavor significance. In: Min DB, Smouse TH, editors. *Flavor chemistry of fats and oils*. Champaign, IL: AOCS; p. 1–37.
- Handa SS, Khanuja SPS, Longo, G, Rakesh, DD. 2008. *Extraction technologies for medicinal and aromatic plants*. Trieste, Italy: United Nations Industrial Development Organization and the International Centre for Science and High Technology (ICS UNIDO).
- Ivanova A, Mikhova B, Najdenski H, Tsvetkova I, Kostova I. 2009. Chemical composition and antimicrobial activity of wild garlic *Allium ursinum* of Bulgarian origin. *Nat Prod Commun.* 4(8):1059–1062.
- Janeczko Z, Sobolewska D. 1995. Czosnek niedźwiedzi – cenna roślina lecznicza/Bear's garlic – a valuable herbal plant. *Wiad. Ziel.* 4:12–14 (in Polish).
- Jimenez-Alvarez D, Giuffrida F, Vanrobaeys F, Golay PA, Cotting C, Lardeau A, Keely BJ. 2008. High-throughput methods to assess lipophilic and hydrophilic antioxidant capacity of food extracts *in vitro*. *J Agric Food Chem.* 56:3470–3477.
- Kozłowska M, Szterk A, Zawada K, Tomasz Z. 2012. New opportunities of the application of natural herb and spice extracts in plant oils: application of electron paramagnetic resonance in examining the oxidative stability. *J Food Sci.* 77(9):994–999.

- Kujala TS, Loponen JM, Klika KD, Pihlaja K. 2000. Phenolics and betacyanins in red beetroot (*Beta vulgaris*) root: distribution and effect of cold storage on the content of total phenolics and three individual Compounds. *J Agric Food Chem.* 48:5338–5342.
- Lante A, Nardi T, Zocca F, Giacomini A, Corich V. 2011. Evaluation of red chicory extract as a natural antioxidant by pure lipid oxidation and yeast oxidative stress response as model systems. *J Agric Food Chem.* 59(10):5318–5324.
- Mendiola JA, Herrero M, Cifuentes A, Ibanez E. 2007. Use of compressed fluids for sample preparation: food applications. *J Chromatogr A.* 1152:234–246.
- Mihaylova D, Bahchevanska S, Toneva V. 2013. Examination of the antioxidant activity of *Haberlea rhodopensis* leaf extracts and their phenolic constituents. *J Food Biochem.* 37(3):255–261.
- Mihaylova D, Lante A, Krastanov A. 2014. Total phenolic content, antioxidant and antimicrobial activity of *Haberlea rhodopensis* extracts obtained by pressurized liquid extraction. *Acta Aliment.* doi:10.1556/AAlim.2014.0009
- Min DB, Boff JM. 2002. Chemistry and reaction of singlet oxygen in foods. *Compr Rev Food Sci Food Saf.* 1:58–72.
- Ou BX, Hampsch-Woodill M, Prior RL. 2001. Development and validation of an improved oxygen radical absorbance capacity assay using fluorescein as the fluorescence probe. *J Agric Food Chem.* 49(10):4619–4626.
- Pârvu M, Pârvu AE, Roșca-Casian O, Vlase L, Groza G. 2010. Antifungal activity of *Allium obliquum*. *J Med Plant Res.* 4:138–141.
- Pârvu M, Toiu A, Vlase L, Pârvu AE. 2010. Determination of some polyphenolic compounds from *Allium* species by HPLC–UV–MS. *Nat Prod Res.* 24:1318–1324.
- Proestos C, Lytoudi K, Mavromelanidou OK, Zoumpoulakis P, Sinanoglou VJ. 2013. Antioxidant capacity of selected plant extracts and their essential oils. *Antioxidants.* 2:11–22, doi:10.3390/antiox2010011.
- Quiroga PR, Grosso NR, Lante A, Lomolino G, Zygadlo JA, Nepote V. 2013. Chemical composition, antioxidant activity and anti-lipase activity of *Origanum vulgare* and *Lippia turbinata* essential oils. *Int J Food Sci Technol.* 48:642–649.
- Re R, Pellegrini N, Proggente A, Pannala A, Yang M, Rice-Evans C. 1999. Antioxidant activity applying an improved ABTS radical cation decolorization assay. *Free Radic Biol Med.* 26(9/10):1231–1237.
- Richter T. 1999. Baerlauch in medizin und mythologie. *Pharm Ztg.* 144:2197–2198.
- Rossetto M, Lante A, Vanzani P, Spettoli P, Scarpa M, Rigo A. 2005. Red chicories as potent scavengers of highly reactive radicals: a study on their phenolic composition and peroxy radical trapping capacity and efficiency. *J Agric Food Chem.* 53:8169–8175.
- Saniewska A, Zuradzka I. 2001. Comparison of fungistatic properties of four varieties of garlic (*Allium sativum* L.) on the example of several species of pathogenic fungi. *Folia Hort.* 13/1A:405–412 (in Polish).
- Sapunjeva T, Alexieva I, Mihaylova D, Popova A. 2012. Antimicrobial and antioxidant activity of extracts of *Allium ursinum* L. *J BioSci Biotechnol, Special edition/Online* :143–145.
- Schmitt B, Schulz H, Storsberg J, Keusgen M. 2005. Chemical characterization of *Allium ursinum* L. depending on harvesting time. *J Agric Food Chem.* 53:7288–7294.
- Sobolewska D, Janeczko Z, Kisiel W, Podolak I, Galanty A, Trojanowska D. 2006. Steroidal glycoside from the underground parts of *Allium ursinum* L., and their cytostatic and antimicrobial activity. *Acta Pol Pharm Drug Res.* 63:219–223.
- Štajner D, Igić R, Popović BM, Malenčić Dj. 2008. Comparative study of antioxidant properties of wild growing and cultivated *Allium* species. *Phytotherapy Res.* 22:113–117.
- Štajner D, Popovic BM, Canadanovic BJ, Stajner M. 2008. Antioxidant and scavenger activities of *Allium ursinum*. *Fitoterapia.* 79(4):303–305.
- Tepe B, Sokmen M, Akpulat HA, Sokmen A. 2006. Screening of the antioxidant potentials of six *Salvia* species from Turkey. *Food Chem.* 95:200–204.
- Wijngaard H, Hossain MB, Rai DK, Brunton N. 2012. Techniques to extract bioactive compounds from food by-products of plant origin. *Food Res Int.* 46:505–513.
- Wu H, Dushenkov S, Ho CT, Sang S. 2009. Novel acetylated flavonoid glycosides from the leaves of *Allium ursinum*. *Food Chem.* 115:592–595.